

ACID -B

- concentration)
- x Small changes of pH from normal can produce marked changes in enzyme activity & chemical reactions within the body
 - x Acidosis -CNS depression, coma if severely acidotic, i.e., pH <7.0)
 - x Alkalosis -CNS excitability, tetany, seizures

$$\text{pH} = \text{pK}_a + \log \frac{[\text{HCO}_3^-]}{[\text{PCO}_2] \cdot 0.03}$$

- x The value of pK_a is 6.1, and the normal HCO_3^- and PCO_2 are 24 mmol/L and 40 mmHg respectively. Substitute these,

$$\begin{aligned} \text{pH} &= 6.1 + \log \frac{24}{40 \cdot 0.03} \\ &= 6.1 + \log 20 \\ &= 6.1 + 1.3 \\ &= 7.4 \end{aligned}$$

Regulation of pH

- x Buffer systems very rapid (seconds), incomplete
- x Respiratory responses -

- x Commonly, the AG remains normal in an acidosis that is due to simple HCO_3^- (as in diarrhea and certain renal diseases) because, as a general principle, it increases to meet the drop in HCO_3^- anions, thereby maintaining anionic balance.
- x When an acid load is present, HCO_3^- excretes acid and anion gap increases, creating a condition known as 'anion gap metabolic acidosis'.

Major Acid Base Disorders and Compensatory Mechanism

Primary Disorder	Primary Disturbance	Primary Compensation
Respiratory Acidosis	$\uparrow \text{PCO}_2$	$\uparrow \text{HCO}_3^-$
Respiratory Alkalosis	$\downarrow \text{PCO}_2$	$\downarrow \text{HCO}_3^-$
Metabolic Acidosis	$\downarrow \text{HCO}_3^-$	$\downarrow \text{PCO}_2$ (hyperventilation)
Metabolic Alkalosis	$\uparrow \text{HCO}_3^-$	$\uparrow \text{PCO}_2$ (hypoventilation)

- x The primary compensation (acute compensation) is generally achieved most rapidly through respiratory control of CO_2
- x Ultimately the renal system excretes acid or bicarbonate (chronic compensation) to reach the final response to the disturbance
- x Mixed disorders are common

Blood Gas Evaluation

What do we get from a blood gas machine?

- x pH - measured
- x PCO_2 - measured
- x PO_2 - measured
- x HCO_3^- - calculated (via Henderson Hesselbalch equation)
- x Base excess (deficit) calculated
- x Hemoglobin oxygen saturation calculated
- x Gases (carbon dioxide & oxygen) are reported as partial pressures only in unit of mmHg (US), or KPa (International); 1 KPa = 7.5 mmHg (torr)

Blood Gas Sampling

- x Arterial vs. venous
- x Venous samples are

Sites for Arterial Samples

- x Dog: dorso pedalfemoral, anterior tibial, lingual
- x Cat: femoral, dorso pedal
- x Horse: facial, mandibular, lateral metatarsal, posterior auricular, carotid
- x Cow: coccygeal, posterior auricular
- x Pig: coccygeal, posterior auricular

Normal Values

	Arterial	Venous
pH	7.35-7.45	7.35
PCO ₂	35-45 mmHg	45 mmHg
PO ₂	90-100 mmHg	40 mmHg
HCO ₃ ⁻		

